

Unsaturated Solution Wikipedia

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SATURATION (CHEMISTRY) - WIKIPEDIA

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ethylene, an unsaturated compound of ethane, ... means that the concentration is actually higher than the saturation point, the solution has become supersaturated.

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define unsaturated. unsaturated synonyms, unsaturated pronunciation, ... "a saturated solution" 2. unsaturated - used of a compound (especially of carbon) ...

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a solution in which the solvent is able to absorb more solute at a particular temperature. want to thank tfd for its existence? tell a friend about us, add a link to ...

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a saturated fat is a type of fat in which the fatty acids all have single bonds. a fat is made of two kinds of smaller molecules: monoglyceride and fatty acids.

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WHAT IS THE DIFFERENCE BETWEEN A UNSATURATED, SATURATED ...

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what is the difference between a unsaturated, saturated, and supersaturated solutions? what is a solution? first of all you should know what a solution is.

UNSATURATED SOLUTION DEFINITION - THOUGHTCO

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unsaturated solution definition. an unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility.

WHAT IS AN UNSATURATED SOLUTION - ANSWERS

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such solution which can dissolve more solute at a given temperature is called as unsaturated solution. a solution that can still dissolve more solute

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unsaturated solution : definition: search for: biology glossary search by everythingbio : a solution in which ...

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so would an unsaturated solution be able to have more solutes dissolved in the solution? ... saturated solution: definition & examples related study materials.

WHAT ARE EXAMPLES OF UNSATURATED SOLUTIONS? | REFERENCE

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unsaturated solutions are solutions that contain less solute than the actual amount of solute that the solvent can dissolve. if more solutes can be dissolved in the ...

UNSATURATED | DEFINE UNSATURATED AT DICTIONARY

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unsaturated definition, not saturated; having the power to dissolve still more of a substance. see more.

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SATURATED SOLUTION | ARTICLE ABOUT SATURATED SOLUTION BY ...

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saturated solution a solution in stable equilibrium with an excess of the solute under given conditions (temperature and pressure). examples are an aqueous solution ...

WHAT DOES "SATURATED SOLUTION" MEAN? | REFERENCE

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what does "saturated solution" mean? a: ... what is an unsaturated solution? a: an unsaturated solution is one that is able to have solvents added or removed.

WHAT IS AN EXAMPLE OF AN UNSATURATED SOLUTION? | YAHOO ANSWERS

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best answer: an unsaturated solution is when there is too little solute in the solvent. when you add salt to water you can mix the solution and the salt ...

CHEMISTRY4GCMS2011 - DIFFERENCE BETWEEN SATURATED ...

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an unsaturated solution is when the solute cannot dissolve into the solvent. for example, if you have a glass of water, ...

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definition of unsaturated solution ... compare with saturated solution .a solution with a concentration lower than its equilibrium solubility .

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saturated solution examples. saturated solution. in chemistry, research into solutions and the dissolving properties of other substances has led to the understanding ...

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